## Notes on the function gsw_alpha_CT_exact(SA,CT,p)

This function, gsw_alpha_CT_exact(SA,CT,p), evaluates the thermal expansion coefficient with respect to constant Conservative Temperature $\Theta, \alpha^{\Theta}$,

$$
\begin{equation*}
\alpha^{\Theta}=-\left.\frac{1}{\rho} \frac{\partial \rho}{\partial \Theta}\right|_{S_{\mathrm{A}}, p}=\left.\frac{1}{v} \frac{\partial v}{\partial \Theta}\right|_{S_{\mathrm{A}}, p} \tag{1}
\end{equation*}
$$

with the input temperature being Conservative Temperature $\Theta$. This function uses the full TEOS-10 Gibbs function $g\left(S_{\mathrm{A}}, t, p\right)$ of IOC et al. (2010), being the sum of the IAPWS-09 and IAPWS-08 Gibbs functions. This function is simply two calls to other GSW functions as follows,

```
t = gsw_t_from_CT(SA,CT,p);
alpha_CT_exact = gsw_alpha_wrt_CT_t_exact(SA,t,p);
```


## References

IAPWS, 2008: Release on the IAPWS Formulation 2008 for the Thermodynamic Properties of Seawater. The International Association for the Properties of Water and Steam. Berlin, Germany, September 2008, available from www.iapws.org. This Release is referred to in the text as IAPWS-08.
IAPWS, 2009: Supplementary Release on a Computationally Efficient Thermodynamic Formulation for Liquid Water for Oceanographic Use. The International Association for the Properties of Water and Steam. Doorwerth, The Netherlands, September 2009, available from http://www.iapws.org. This Release is referred to in the text as IAPWS-09.
IOC, SCOR and IAPSO, 2010: The international thermodynamic equation of seawater - 2010: Calculation and use of thermodynamic properties. Intergovernmental Oceanographic Commission, Manuals and Guides No. 56, UNESCO (English), 196 pp. Available from http://www.TEOS-10.org

Here follows section 2.18 and appendix A. 15 of the TEOS-10 Manual (IOC et al., 2010).

### 2.18 Thermal expansion coefficients

The thermal expansion coefficient $\alpha^{t}$ with respect to in situ temperature $t$, is

$$
\begin{equation*}
\alpha^{t}=\alpha^{t}\left(S_{\mathrm{A}}, t, p\right)=-\left.\frac{1}{\rho} \frac{\partial \rho}{\partial T}\right|_{S_{\mathrm{A}}, p}=\left.\frac{1}{v} \frac{\partial v}{\partial T}\right|_{S_{\mathrm{A}}, p}=\frac{g_{T P}}{g_{P}} \tag{2.18.1}
\end{equation*}
$$

The thermal expansion coefficient $\alpha^{\theta}$ with respect to potential temperature $\theta$, is (see appendix A.15)

$$
\begin{equation*}
\alpha^{\theta}=\alpha^{\theta}\left(S_{\mathrm{A}}, t, p, p_{\mathrm{r}}\right)=-\left.\frac{1}{\rho} \frac{\partial \rho}{\partial \theta}\right|_{S_{\mathrm{A}}, p}=\left.\frac{1}{v} \frac{\partial v}{\partial \theta}\right|_{S_{\mathrm{A}}, p}=\frac{g_{T P}}{g_{P}} \frac{g_{T T}\left(S_{\mathrm{A}}, \theta, p_{\mathrm{r}}\right)}{g_{T T}}, \tag{2.18.2}
\end{equation*}
$$

where $p_{\mathrm{r}}$ is the reference pressure of the potential temperature. The $g_{T T}$ derivative in the numerator is evaluated at $\left(S_{\mathrm{A}}, \theta, p_{\mathrm{r}}\right)$ whereas the other derivatives are all evaluated at $\left(S_{\mathrm{A}}, t, p\right)$.

The thermal expansion coefficient $\alpha^{\Theta}$ with respect to Conservative Temperature $\Theta$, is (see appendix A.15)

$$
\begin{equation*}
\alpha^{\Theta}=\alpha^{\Theta}\left(S_{\mathrm{A}}, t, p\right)=-\left.\frac{1}{\rho} \frac{\partial \rho}{\partial \Theta}\right|_{S_{\mathrm{A}}, p}=\left.\frac{1}{v} \frac{\partial v}{\partial \Theta}\right|_{S_{\mathrm{A}}, p}=-\frac{g_{T P}}{g_{P}} \frac{c_{p}^{0}}{\left(T_{0}+\theta\right) g_{T T}} . \tag{2.18.3}
\end{equation*}
$$

Note that Conservative Temperature $\Theta$ is defined only with respect to a reference pressure of 0 dbar so that the $\theta$ in Eqn. (2.18.3) is the potential temperature with $p_{r}=0$ dbar. All the derivatives on the right-hand side of Eqn. (2.18.3) are evaluated at ( $S_{\mathrm{A}}, t, p$ ). The constant $c_{p}^{0}$ is defined in Eqn. (3.3.3) below.

## A. 15 Derivation of the expressions for $\alpha^{\theta}, \beta^{\theta}, \alpha^{\Theta}$ and $\beta^{\Theta}$

This appendix derives the expressions in Eqns. (2.18.2) - (2.18.3) and (2.19.2) - (2.19.3) for the thermal expansion coefficients $\alpha^{\theta}$ and $\alpha^{\Theta}$ and the haline contraction coefficients $\beta^{\theta}$ and $\beta^{\ominus}$.

In order to derive Eqn. (2.18.2) for $\alpha^{\theta}$ we first need an expression for $\partial \theta /\left.\partial T\right|_{S_{A}, p}$. This is found by differentiating with respect to in situ temperature the entropy equality $\eta\left(S_{\mathrm{A}}, t, p\right)=\eta\left(\mathrm{S}_{\mathrm{A}}, \theta\left[\mathrm{S}_{\mathrm{A}}, t, p, p_{\mathrm{r}}\right], p_{\mathrm{r}}\right)$ which defines potential temperature, obtaining

$$
\begin{equation*}
\left.\frac{\partial \theta}{\partial T}\right|_{S_{\mathrm{A}}, p}=\frac{\eta_{T}\left(S_{\mathrm{A}}, t, p\right)}{\eta_{T}\left(S_{\mathrm{A}}, \theta, p_{\mathrm{r}}\right)}=\frac{g_{T T}\left(S_{\mathrm{A}}, t, p\right)}{g_{T T}\left(S_{\mathrm{A}}, \theta, p_{\mathrm{r}}\right)}=\frac{\left(T_{0}+\theta\right)}{\left(T_{0}+t\right)} \frac{c_{p}\left(S_{\mathrm{A}}, t, p\right)}{c_{p}\left(S_{\mathrm{A}}, \theta, p_{\mathrm{r}}\right)} . \tag{A.15.1}
\end{equation*}
$$

This is then used to obtain the desired expression Eqn. (2.18.2) for $\alpha^{\theta}$ as follows

$$
\begin{equation*}
\alpha^{\theta}=\left.\frac{1}{v} \frac{\partial v}{\partial \theta}\right|_{S_{\mathrm{A}}, p}=\left.\frac{1}{v} \frac{\partial v}{\partial T}\right|_{S_{\mathrm{A}}, p}\left(\left.\frac{\partial \theta}{\partial T}\right|_{S_{\mathrm{A}}, p}\right)^{-1}=\frac{g_{T P}\left(S_{\mathrm{A}}, t, p\right)}{g_{P}\left(S_{\mathrm{A}}, t, p\right)} \frac{g_{T T}\left(S_{\mathrm{A}}, \theta, p_{\mathrm{r}}\right)}{g_{T T}\left(S_{\mathrm{A}}, t, p\right)} . \tag{A.15.2}
\end{equation*}
$$

In order to derive Eqn. (2.18.3) for $\alpha^{\Theta}$ we first need an expression for $\partial \Theta /\left.\partial t\right|_{S_{A}, p}$. This is found by differentiating with respect to in situ temperature the entropy equality $\eta\left(S_{\mathrm{A}}, t, p\right)=\hat{\eta}\left(S_{\mathrm{A}}, \Theta\left[S_{\mathrm{A}}, t, p\right]\right)$ obtaining

$$
\begin{equation*}
\left.\frac{\partial \Theta}{\partial T}\right|_{S_{\mathrm{A}}, p}=\left.\eta_{T}\left(S_{\mathrm{A}}, t, p\right) \frac{\partial \Theta}{\partial \eta}\right|_{S_{\mathrm{A}}}=-\left(T_{0}+\theta\right) \frac{g_{T T}\left(S_{\mathrm{A}}, t, p\right)}{c_{p}^{0}}=\frac{\left(T_{0}+\theta\right)}{\left(T_{0}+t\right)} \frac{c_{p}\left(S_{\mathrm{A}}, t, p\right)}{c_{p}^{0}}, \tag{A.15.3}
\end{equation*}
$$

where the second part of this equation has used Eqn. (A.12.4) for $\left.\Theta_{\eta}\right|_{S_{A}}$. This is then used to obtain the desired expression Eqn. (2.18.3) for $\alpha^{\Theta}$ as follows

$$
\begin{equation*}
\alpha^{\Theta}=\left.\frac{1}{v} \frac{\partial v}{\partial \Theta}\right|_{S_{A}, p}=\left.\frac{1}{v} \frac{\partial v}{\partial T}\right|_{S_{A}, p}\left(\left.\frac{\partial \Theta}{\partial T}\right|_{S_{A}, p}\right)^{-1}=-\frac{g_{T P}\left(S_{\mathrm{A}}, t, p\right)}{g_{P}\left(S_{\mathrm{A}}, t, p\right)} \frac{c_{p}^{0}}{\left(T_{0}+\theta\right) g_{T T}\left(S_{\mathrm{A}}, t, p\right)} . \tag{A.15.4}
\end{equation*}
$$

In order to derive Eqn. (2.19.2) for $\beta^{\theta}$ we first need an expression for $\partial \theta /\left.\partial S_{\mathrm{A}}\right|_{T, p}$. This is found by differentiating with respect to Absolute Salinity the entropy equality $\eta\left(S_{\mathrm{A}}, t, p\right)=\eta\left(S_{\mathrm{A}}, \theta\left[S_{\mathrm{A}}, t, p, p_{\mathrm{r}}\right], p_{\mathrm{r}}\right)$ which defines potential temperature, obtaining

$$
\begin{align*}
\left.\frac{\partial \theta}{\partial S_{\mathrm{A}}}\right|_{T, p} & =\left.\theta_{\eta}\right|_{S_{\mathrm{A}}}\left[\eta_{S_{\mathrm{A}}}\left(S_{\mathrm{A}}, t, p\right)-\eta_{S_{\mathrm{A}}}\left(S_{\mathrm{A}}, \theta, p_{\mathrm{r}}\right)\right] \\
& =\frac{\left(T_{0}+\theta\right)}{c_{p}\left(S_{\mathrm{A}}, \theta, p_{\mathrm{r}}\right)}\left[\mu_{T}\left(S_{\mathrm{A}}, \theta, p_{\mathrm{r}}\right)-\mu_{T}\left(S_{\mathrm{A}}, t, p\right)\right]  \tag{A.15.5}\\
& =\left[g_{S_{\mathrm{A}} T}\left(S_{\mathrm{A}}, t, p\right)-g_{S_{\mathrm{A}} T}\left(S_{\mathrm{A}}, \theta, p_{\mathrm{r}}\right)\right] / g_{T T}\left(S_{\mathrm{A}}, \theta, p_{\mathrm{r}}\right),
\end{align*}
$$

where Eqns. (A.12.5) and (A.12.7) have been used with a general reference pressure $p_{\mathrm{r}}$ rather than with $p_{\mathrm{r}}=0$. By differentiating $\rho=\tilde{\rho}\left(S_{\mathrm{A}}, \theta\left[S_{\mathrm{A}}, t, p, p_{\mathrm{r}}\right], p\right)$ with respect to Absolute Salinity it can be shown that (Gill (1982), McDougall (1987a))

$$
\begin{equation*}
\beta^{\theta}=\left.\frac{1}{\rho} \frac{\partial \rho}{\partial S_{\mathrm{A}}}\right|_{\theta, p}=\left.\frac{1}{\rho} \frac{\partial \rho}{\partial S_{\mathrm{A}}}\right|_{T, p}+\left.\alpha^{\theta} \frac{\partial \theta}{\partial S_{\mathrm{A}}}\right|_{T, p}, \tag{A.15.6}
\end{equation*}
$$

and using Eqn. (A.15.5) we arrive at the desired expression Eqn. (2.19.2) for $\beta^{\theta}$

$$
\begin{equation*}
\beta^{\theta}=-\frac{g_{S_{\mathrm{A}} P}\left(S_{\mathrm{A}}, t, p\right)}{g_{P}\left(S_{\mathrm{A}}, t, p\right)}+\frac{g_{T P}\left(S_{\mathrm{A}}, t, p\right)\left[g_{S_{\mathrm{A}} T}\left(S_{\mathrm{A}}, t, p\right)-g_{S_{\mathrm{A}} T}\left(S_{\mathrm{A}}, \theta, p_{\mathrm{r}}\right)\right]}{g_{P}\left(S_{\mathrm{A}}, t, p\right) g_{T T}\left(S_{\mathrm{A}}, t, p\right)} . \tag{A.15.7}
\end{equation*}
$$

Note that the terms in the natural logarithm of the square root of Absolute Salinity cancel from the two parts of the square brackets in Eqns. (A.15.5) and (A.15.7).

In order to derive Eqn. (2.19.3) for $\beta^{\Theta}$ we first need an expression for $\partial \Theta /\left.\partial S_{\mathrm{A}}\right|_{T, p}$. This is found by differentiating with respect to Absolute Salinity the entropy equality $\eta\left(S_{\mathrm{A}}, t, p\right)=\hat{\eta}\left(S_{\mathrm{A}}, \Theta\left[S_{\mathrm{A}}, t, p\right]\right)$ obtaining (using Eqns. (A.12.4) and (A.12.8))

$$
\begin{align*}
\left.\frac{\partial \Theta}{\partial S_{\mathrm{A}}}\right|_{T, p} & =\left.\Theta_{\eta}\right|_{S_{\mathrm{A}}}\left[\eta_{S_{\mathrm{A}}}\left(S_{\mathrm{A}}, t, p\right)-\left.\hat{\eta}_{S_{\mathrm{A}}}\right|_{\Theta}\right] \\
& =\left[\mu\left(S_{\mathrm{A}}, \theta, 0\right)-\left(T_{0}+\theta\right) \mu_{T}\left(S_{\mathrm{A}}, t, p\right)\right] / c_{p}^{0}  \tag{A.15.8}\\
& =\left[g_{S_{\mathrm{A}}}\left(S_{\mathrm{A}}, \theta, 0\right)-\left(T_{0}+\theta\right) g_{S_{\mathrm{A}} T}\left(S_{\mathrm{A}}, t, p\right)\right] / c_{p}^{0}
\end{align*}
$$

Differentiating $\rho=\hat{\rho}\left(S_{\mathrm{A}}, \Theta\left[S_{\mathrm{A}}, t, p\right], p\right)$ with respect to Absolute Salinity leads to

$$
\begin{equation*}
\beta^{\Theta}=\left.\frac{1}{\rho} \frac{\partial \rho}{\partial S_{\mathrm{A}}}\right|_{\Theta, p}=\left.\frac{1}{\rho} \frac{\partial \rho}{\partial S_{\mathrm{A}}}\right|_{T, p}+\left.\alpha^{\Theta} \frac{\partial \Theta}{\partial S_{\mathrm{A}}}\right|_{T, p}, \tag{A.15.9}
\end{equation*}
$$

and using Eqn. (A.15.8) we arrive at the desired expression (2.19.3) for $\beta^{\Theta}$ namely

$$
\begin{equation*}
\beta^{\Theta}=-\frac{g_{S_{\mathrm{A}} P}\left(S_{\mathrm{A}}, t, p\right)}{g_{P}\left(S_{\mathrm{A}}, t, p\right)}+\frac{g_{T P}\left(S_{\mathrm{A}}, t, p\right)\left[g_{S_{\mathrm{A}} T}\left(S_{\mathrm{A}}, t, p\right)-g_{S_{\mathrm{A}}}\left(S_{\mathrm{A}}, \theta, 0\right) /\left(T_{0}+\theta\right)\right]}{g_{P}\left(S_{\mathrm{A}}, t, p\right) g_{T T}\left(S_{\mathrm{A}}, t, p\right)} . \tag{A.15.10}
\end{equation*}
$$

Note that the terms in the natural logarithm of the square root of Absolute Salinity cancel from the two parts of the square brackets in Eqns. (A.15.8) and (A.15.10).

